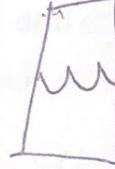
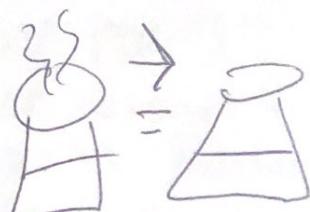


SI session: Chapter 3

1. Define and give an example for the following terms:

Physical Properties	<ul style="list-style-type: none"> Properties of a substance that can identify the substance w/o causing a change in the composition. 	<ul style="list-style-type: none"> Color malleability taste density ductility.
Chemical Properties	<ul style="list-style-type: none"> Properties of a substance that relate to changes in the composition of the substance or its reactions w/ other substances. 	<ul style="list-style-type: none"> -Flamability- -Oxidation- (rust).
Extensive properties Physical Change.	<ul style="list-style-type: none"> Properties that relate to how much of a substance is present. 	 <ul style="list-style-type: none"> MASS / weight.
Intensive properties Chem- Change.	<ul style="list-style-type: none"> Properties that relate to a substance's identity. 	<ul style="list-style-type: none"> Flamability rust
Law of conservation and mass	<ul style="list-style-type: none"> States that mass is neither created nor destroyed. 	
Law of Conservation of energy	<ul style="list-style-type: none"> States that energy is not created nor destroyed. 	<ul style="list-style-type: none"> Constant.

Energy	<ul style="list-style-type: none"> The capacity to do work 	

2. Describe the difference between a physical and chemical change.

Physical Change - Characteristic properties not altered - Change in extensive Property.

Chemical Change - Alters characteristic Properties, New Matter type formed - intensive property.

3. Describe the difference between exothermic and endothermic. Change.

Exothermic - Releases heat - get warmer.

Endo - absorbs heat - get colder
Cold Packs.

4. Name and define the types of energy.

- Kinetic - energy associated w/ motion
- Potential - \downarrow w/ position
- Thermal - random motions of atoms + molecules in matter
- Electrical - flow of electrons.

5. Describe the Fahrenheit scale.

penn
GEAR

Freezing pt. -32°F

Boiling pt. -212°F

$$\cancel{1.24 \times 10^3}$$

Review of math:

$$\begin{array}{c} M \\ \diagdown \\ D \\ \diagup \\ V \end{array}$$

mass

volume

- Calculate the volume of an object with a mass of 0.236 cm^3 and a density of 5.26 g/cm^3 . provide your answer in Kilograms.

$$M = VD$$

$$M = (0.236 \text{ cm}^3) (5.26 \text{ g/cm}^3)$$
$$M = 1.24 \text{ g}$$

$$\begin{array}{c} 1.24 \text{ g} \\ \hline | & 1 \text{ kg} \\ | & 1000 \text{ g} \end{array} = 0.00124$$
$$1.24 \times 10^{-3}$$

- A ball rolls down a hill at a speed of 5 miles per hour. If the ball travels for 30 meters, how many centimeters did the ball travel?

$$\begin{array}{c} 30 \text{ meter} \\ \hline | & 100 \text{ cm} \\ | & 1 \text{ meters} \end{array} = 3000 \text{ cm}$$
$$3.0 \times 10^3$$