## Chapter 5 Continued:

1. Define and provide an example for the following terms:

Term	Definition	Example
Polyatomic ions		·
Acids		
Binary Acids		
Oxyacids		
Formula Mass		
2 Molocular Compound Nom		

2. Molecular Compound Nomenclature problems:

a. CO2

	c. Silicon dioxide
	d. SeBr2
	e. Silicon tetraiodide
3.	What are the rules for naming acids?
4.	What are the rules for naming polyatomic compounds?

b. NCl3



a. HBrO4

b. HI

c. HIO4

d. Hydrobromic acid

e. Arsenous acid

6. Solve the formula mass for KIO4